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Algebra 1: Common Core (15th Edition) Charles, Randall I. Publisher Prentice Hall ISBN 978-0-13328-114-9

Stability constants are fundamental to understanding the behavior of metal ions in aqueous solution. Such understanding is important in a wide variety of areas, such as metal ions in biology, biomedical applications, metal ions in the environment, extraction metallurgy, food chemistry, and metal ions in many industrial processes. In

spite of this importance, it appears that many inorganic chemists have lost an appreciation for the importance of stability constants, and the thermodynamic aspects of complex formation, with attention focused over the last thirty years on newer areas, such as organometallic chemistry. This book is an attempt to show the richness of chemistry that can be revealed by stability constants, when measured as part of an overall strategy aimed at understanding the complexing properties of a particular ligand or metal ion. Thus, for example, there are numerous crystal structures of the  $\text{Li}^+$  ion with crown ethers. What do these indicate to us about the chemistry of  $\text{Li}^+$  with crown ethers? In fact, most of these crystal structures are in a sense misleading, in that the  $\text{Li}^+$  ion forms no complexes, or at best very weak complexes, with familiar crown ethers such as 12-crown-4, in any known solvent. Thus, without the stability constants, our understanding of the chemistry of a metal ion with any particular ligand must be regarded as incomplete. In this book we attempt to show how stability constants can reveal factors in ligand design which could not readily be deduced from any other physical technique.

NOTE: This edition features the same content as the traditional text in a convenient, three-hole-punched, loose-leaf version. Books a la Carte also offer a great value--this format costs significantly less than a new textbook. Before purchasing, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of Pearson's MyLab & Mastering products exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, you may need a CourseID, provided by your instructor, to register for and use Pearson's MyLab & Mastering products. xxxxxxxxxxxxxxxx For two-semester general chemistry courses (science majors). Make critical connections in chemistry clear and visible McMurry/Fay/Robinson's Chemistry, Seventh Edition, aims to help students understand the connections between topics in general chemistry and why they matter. The Seventh Edition provides a concise and streamlined narrative that blends the quantitative and visual aspects of chemistry, demonstrates the connections between topics, and illustrates the application of chemistry to their lives and careers. New content offers a better bridge between organic and biochemistry and general chemistry content, and new and improved pedagogical features make the text a true teaching tool rather than just a reference book. New MasteringChemistry features include conceptual worked examples and integrated Inquiry sections that help make critical connections clear and visible and increase students' understanding of chemistry. The Seventh Edition fully integrates the text with new MasteringChemistry content and functionality to support the learning process before, during, and after class. Also Available with MasteringChemistry®. MasteringChemistry from Pearson is the leading online homework, tutorial, and assessment system, designed to improve results by engaging students before, during, and after class with powerful content. Instructors ensure students arrive ready to learn by assigning educationally effective content before class, and encourage critical thinking and retention with in-class resources such as Learning Catalytics. Students can further master concepts after class through traditional and adaptive homework assignments that provide hints and answer-specific feedback. The Mastering gradebook records scores for all automatically graded assignments in one place, while diagnostic tools give instructors access to rich data to assess student understanding and misconceptions. Mastering brings learning full circle by continuously adapting to each student and making learning more personal than ever-before, during, and after class.

Volume 6 of the successful series 'Reviews in Computational Chemistry' contains articles of interest to pharmaceutical chemists, biological chemists, chemical engineers, inorganic and organometallic chemists, synthetic organic chemists, polymer chemists, and theoretical chemists. The series is designed to help the chemistry community keep current with the many new developments in computational techniques. The writing style is refreshingly pedagogical and non-mathematical, allowing students and researchers access to computational methods outside their immediate area of expertise.

"A learner-centered approach means the book should recognize that all students come to this course with different goals and aspirations, different backgrounds, different experiences, and different plans for learning. Accordingly, this textbook affirms that organic chemistry [and science in general] is for everyone, regardless of background, and that all are welcome and supported here. I wanted the book to happen the way a conversation about organic chemistry would happen, allowing for it to spin off organically [Ha!] in a variety of directions. The book is not intended to be a lecture from a podium to an unengaged audience. Instead, it is a lively discussion, a back and forth between student and text, the same kind of discussion that might occur in office hours, in the university cafeteria, while standing over a reaction in chemistry lab, or during a late-night review session"--

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