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a. calorimeter

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unit of energy

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$2\text{Al}_2\text{O}_3$ (s)

A) exothermic,
released

B) exothermic,
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Answer: Change
in internal
energy (ΔU) for
an isolated
system is zero
for it does not

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exchange any
energy with the
surroundings.

But entropy
tends to
increase in case
of spontaneous
reaction.

Therefore, $\Delta S >$
0 or positive.

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said to be _____
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H at constant
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According to
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